

## Progress Test Files 1–6

### Grammar, Vocabulary, and Pronunciation



#### 5 Write the opposite.

Example: high *low*

- |                   |                   |
|-------------------|-------------------|
| 1 strong _____    | 6 difficult _____ |
| 2 always _____    | 7 safe _____      |
| 3 cheap _____     | 8 get up _____    |
| 4 dirty _____     | 9 turn on _____   |
| 5 beautiful _____ | 10 start _____    |

10

#### 6 Complete the words in the sentences.

Example: I usually ~~sleep~~ for eight hours every night.

- They want to g\_\_\_\_\_ married in July.
- Can you s\_\_\_\_\_ another language?
- I e\_\_\_\_\_ my emails five times a day.
- My sister's an administrator. She works in an o\_\_\_\_\_.
- I usually w\_\_\_\_\_ up very late at the weekend.
- Jackie's son loves planes. He wants to be a p\_\_\_\_\_.
- I r\_\_\_\_\_ a newspaper every weekend.
- Do you l\_\_\_\_\_ in a house or a flat?
- I always h\_\_\_\_\_ breakfast before I go to work.
- Helen is tall, but her brother is s\_\_\_\_\_.

10

#### 7 Complete the sentences with the correct word.

Example: I usually pay in shops with my ~~credit card~~.  
stamps credit card identity card

- Let's \_\_\_\_\_ shopping on Saturday.  
do go get
- My son likes \_\_\_\_\_ to the radio.  
having hearing listening
- Her mum's a \_\_\_\_\_. She works in a hospital.  
housewife nurse journalist
- I'd like to be a \_\_\_\_\_. I love playing the guitar.  
musician painter music
- What do you usually do \_\_\_\_\_ the weekend?  
by at to
- Rob \_\_\_\_\_ a lot of exercise. He cycles to work every day.  
practises makes does
- I usually go \_\_\_\_\_ at 8.00.  
home to home at home
- Oliver's in the kitchen. He's \_\_\_\_\_ the dinner.  
having making taking

- My phone doesn't \_\_\_\_\_.  
work go speak
- I can't \_\_\_\_\_ the housework now because I'm very tired.  
make have do

10

Vocabulary total  40

#### PRONUNCIATION

#### 8 Underline the stressed syllable.

Example: g|range

- |               |                |
|---------------|----------------|
| 1 ex pen sive | 6 twen ti eth  |
| 2 ti ssue     | 7 e mail       |
| 3 news pa per | 8 grand mother |
| 4 ja pan      | 9 i ta lian    |
| 5 home work   | 10 Wednes day  |

10

#### 9 Match the words with the same sounds.

find file him tall open cousin glasses  
receptionist shopping June house

Example: shower *house*

- |                  |
|------------------|
| 1 German _____   |
| 2 horse _____    |
| 3 photo _____    |
| 4 clock _____    |
| 5 she _____      |
| 6 diary _____    |
| 7 fish _____     |
| 8 scissors _____ |
| 9 flower _____   |
| 10 jazz _____    |

10

Pronunciation total  20

Grammar, Vocabulary, and Pronunciation total  100

# Name Class English File End Of Course Test Grammar

**Arturo Cuomo**



## **Name Class English File End Of Course Test Grammar:**

English File 4E Pre-intermediate Student Book Christina Latham-Koenig, Clive Oxenden, Kate Chomacki, Jerry Lambert, 2020-07-28 English File's unique lively and enjoyable lessons are renowned for getting students talking. In fact 90% of English File teachers we surveyed in our impact study found that the course improves students' speaking skills.

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction  $2\text{Ag}^+ + \text{CrO}_4^{2-} \rightarrow \text{Ag}_2\text{CrO}_4$

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web jan 29 2020 the ionic form of this reaction is  $2\text{Na}^+ + \text{Cl}^- + \text{Cu}^{2+} + \text{SO}_4^{2-} \rightarrow \text{CuSO}_4 + 2\text{Na}^+ + \text{SO}_4^{2-}$  the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web the six  $\text{NO}_3^-$  ions and the six  $\text{Na}^+$  ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation  $3\text{Ba}^{2+} + 2\text{PO}_4^{3-} \rightarrow \text{Ba}_3(\text{PO}_4)_2$

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web these spectator ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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web practice problems on net ionic equations page 2 of 3 answer key to practice problems on net ionic equations 1 molecular  $\text{AgNO}_3 + \text{KCl} \rightarrow \text{AgCl} + \text{KNO}_3$  total ionic  $\text{Ag}^+ + \text{NO}_3^- + \text{K}^+ + \text{Cl}^- \rightarrow \text{AgCl} + \text{K}^+ + \text{NO}_3^-$  net ionic  $\text{Ag}^+ + \text{Cl}^- \rightarrow \text{AgCl}$

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web video answer you know spectator ions are the ions present on the both sides of the equation indicate cancels out in the total ionic equations so you know in a case chemical reactions are described with chemical equations which symbolically present the starting materials end products

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web net ionic equations an equation for a reaction in solution showing only those particles that are directly involved in the chemical change spectator ions an ion that is not directly involved in a chemical reaction an ion that does not change

oxidation number or composition during a reaction products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water h aq oh aq h  
2 o l the spectator ions in koh aq and hno 3 aq are k aq and no 3 aq how to identify and list the

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spectator ion can therefore be observed in the reaction of aqueous solutions of sodium carbonate and copper ii sulfate but  
does not affect the equilibrium what are spectator ions

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strong electrolytes shown as ions

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two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both  
before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and  
ions that are directly involved in the chemical reaction

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spectator ions for three different sets of reactants and to answer a short question requiring no calculator about each reaction

the questions are intended to test the students understanding of the meaning of the reactions

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web jun 2 2023 the spectator ions are cut out from a reaction hence the net ionic equation for the above reaction would be  $2\text{Cl}^- + \text{Cu}^{2+} + 2\text{Cl}^- + \text{S} + 2\text{Cl}^- + \text{Cu}^{2+} + 2\text{Cl}^- + \text{S}$  these spectator ions are not actively involved in the reaction the existence of spectator ions affects the Debye length

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