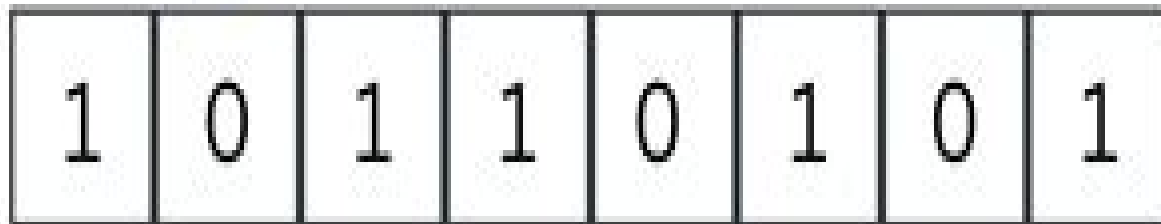


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web mar 6 2017 1 the chemical reaction $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$ is a synthesis reaction decomposition reaction single displacement reaction double displacement reaction combustion reaction you can recognize a decomposition or analysis reaction by noticing a compound or molecule breaks into smaller chemical species 2 the chemical reaction

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web learn test match q chat created by meathello terms in this set 21 the atomic mass of nitrogen is 14.01 hydrogen is 1.01 sulfur is 32.07 and oxygen is 16.00 what is the molar mass of ammonium sulphate NH_4SO_4 137.17 g the atomic mass of carbon is 12.01 sodium is 22.99 and oxygen is 16.00

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web for each reaction in part i and part ii record your observations molecular equation total ionic equation and net ionic

equation make sure to write any evidence of a chemical reaction with sufficient detail to help you

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web what type of chemical reaction is shown $\text{AB} + \text{CD} \rightarrow \text{AD} + \text{BC}$ decomposition the reaction $\text{Pb} + \text{NO}_3 \rightarrow 2\text{KI} + \text{PbI}_2 + \text{KNO}_3$ is a double replacement reaction the reaction $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ is a synthesis

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