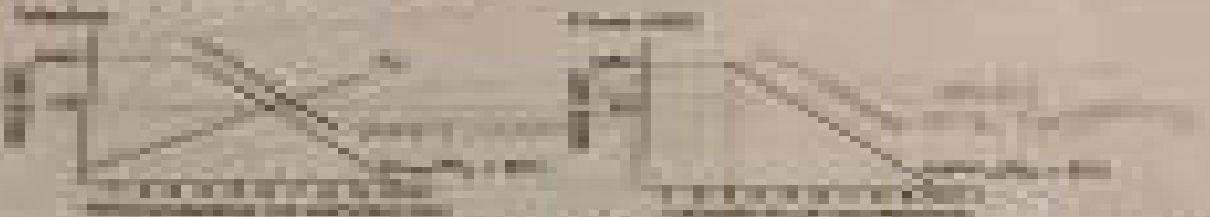


## 1.1 Summary of the adjustment in IS-LM model

Figure 1.1 illustrates the adjustment process in the IS-LM model. The initial equilibrium is at the intersection of the IS curve and the LM curve, corresponding to an interest rate  $r_0$  and output  $Y_0$ . A decrease in the money supply shifts the LM curve upwards to  $LM_1$ , leading to a new equilibrium at a higher interest rate  $r_1$  and lower output  $Y_1$ . The adjustment process is shown by the sequence of points  $A, B, C, D, E, F, G, H, I, J, K, L, M, N, O, P, Q, R, S, T, U, V, W, X, Y, Z$  as the economy moves towards the new equilibrium.

### 1.2 IS-LM model and adjustment in the IS-LM model



1. The IS-LM model is a macroeconomic model that describes the relationship between the real interest rate and the level of output in the economy.
2. The IS curve represents the goods market equilibrium, and the LM curve represents the money market equilibrium.
3. The IS-LM model is used to analyze the effects of monetary and fiscal policy on the economy.
4. The IS-LM model is a simplified version of the more complex macroeconomic models.
5. The IS-LM model is a static model, meaning it does not take into account the dynamic aspects of the economy.
6. The IS-LM model is a partial equilibrium model, meaning it only considers the relationship between the interest rate and output, and does not take into account other aspects of the economy.
7. The IS-LM model is a macroeconomic model that describes the relationship between the real interest rate and the level of output in the economy.
8. The IS-LM model is used to analyze the effects of monetary and fiscal policy on the economy.
9. The IS-LM model is a simplified version of the more complex macroeconomic models.
10. The IS-LM model is a static model, meaning it does not take into account the dynamic aspects of the economy.
11. The IS-LM model is a partial equilibrium model, meaning it only considers the relationship between the interest rate and output, and does not take into account other aspects of the economy.

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho/para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno<sub>3</sub> 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion  $\text{NO}_2^+$  which acts as the electrophilic species

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