

**0 9** . **4** In an experiment starting with 5.05 g of phenylamine, 4.82 g of purified product were obtained in step 1.

Calculate the percentage yield in this reaction.  
Give your answer to the appropriate number of significant figures.

**[3 marks]**

Percentage yield = \_\_\_\_\_ %

**0 9** . **5** A reagent for step 2 is a mixture of concentrated nitric acid and concentrated sulfuric acid, which react together to form a reactive intermediate.

Write an equation for the reaction of this intermediate in step 2.

**[1 mark]**

\_\_\_\_\_

**0 9** . **6** Name a mechanism for the reaction in step 2.

**[1 mark]**

\_\_\_\_\_

**0 9** . **7** Suggest the type of reaction occurring in step 3.

**[1 mark]**

\_\_\_\_\_

**0 9** . **8** Identify the reagents used in step 4.

**[1 mark]**

\_\_\_\_\_

# Aqa S Specimen Paper 2

**BM King**



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