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# BASIC ELECTRICAL ENGINEERING



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web in the context of a chemical reaction the term neutralization is used for a reaction between an acid and a base or alkali historically this reaction was represented as acid base alkali salt water for example  $\text{hcl naoh nacl h}_2\text{o}$

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web during this reaction the  $\text{h}^+$  ions from the acid and the  $\text{oh}^-$  ions from the alkali react together to form water as this happens energy is released which is detected by the thermometer

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable  
measure an appropriate volume of each liquid eg 25 cm<sup>3</sup> place one of the  
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exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of  
sodium hydroxide solution with HCl as 57.9 kJ mol<sup>-1</sup>  
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produce salts neutralisation is the reaction between an acid and a base  
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eg 1 HCl(aq) + NaOH(aq) → NaCl(aq) + H<sub>2</sub>O(l) for which ΔH = 57 kJ mol<sup>-1</sup> eg 2 HNO<sub>3</sub>(aq) + KOH(aq) → KNO<sub>3</sub>(aq) + H<sub>2</sub>O(l) for which ΔH = 57 kJ mol<sup>-1</sup>  
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together to produce one mole of water for example HCl(aq) + NaOH(aq) → NaCl(aq) + H<sub>2</sub>O(aq) you can see that one mole of water is  
produced when one mole of HCl or one mole of NaOH reacts

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alfred enoch who also acted in the harry potter movie franchise

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