

CHAPTER 4 REVIEW

Chemical Equations and Reactions

Short Review

Write a balanced chemical equation for the reaction described.

1. _____
 - a. balanced chemical equation represents all the following except:
 - i an experimentally established law
 - ii substances by which reactions combine to form products
 - iii identities of reactant and products in a chemical reaction
 - iv relative quantities of reactants and products in a chemical reaction
 - b. According to the law of conservation of mass, the total mass of the
 - i reactants is
 - i the same as the total mass of the products
 - ii always less than total mass of the products
 - iii sometimes more than total mass of the products
 - iv always greater than total mass of the products
2. Write a balanced chemical equation, structural formulas will count. For each reaction there will be one. Show the reaction type and complete the chemical equation leaving in the products. Minimum balancing the final equation. Dimensionality only can be Figure 4.3 in the text.
 - a. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$



Chapter 8 Chemical Reactions Guided Reading Answers

Clifford Lane



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