

## Section 12.1

### OBJECTIVE 1

Identify suitable Lewis acid-base adducts from the species provided:

1.  $\text{BF}_3$ 
  - (a)  $\text{NH}_3$
  - (b)  $\text{H}_2\text{O}$
  - (c)  $\text{CH}_3\text{OH}$
  - (d)  $\text{C}_2\text{H}_6$
  - (e)  $\text{C}_2\text{H}_4$
  - (f)  $\text{C}_2\text{H}_2$
  - (g)  $\text{C}_2\text{H}_6\text{O}$
  - (h)  $\text{C}_2\text{H}_6\text{S}$
  - (i)  $\text{C}_2\text{H}_6\text{N}_2$
  - (j)  $\text{C}_2\text{H}_6\text{N}_2\text{O}$
  - (k)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_2$
  - (l)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_3$
  - (m)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_4$
  - (n)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_5$
  - (o)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_6$
  - (p)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_7$
  - (q)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_8$
  - (r)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_9$
  - (s)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{10}$
  - (t)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{11}$
  - (u)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{12}$
  - (v)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{13}$
  - (w)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{14}$
  - (x)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{15}$
  - (y)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{16}$
  - (z)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{17}$
  - (aa)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{18}$
  - (ab)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{19}$
  - (ac)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{20}$
  - (ad)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{21}$
  - (ae)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{22}$
  - (af)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{23}$
  - (ag)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{24}$
  - (ah)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{25}$
  - (ai)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{26}$
  - (aj)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{27}$
  - (ak)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{28}$
  - (al)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{29}$
  - (am)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{30}$
  - (an)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{31}$
  - (ao)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{32}$
  - (ap)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{33}$
  - (aq)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{34}$
  - (ar)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{35}$
  - (as)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{36}$
  - (at)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{37}$
  - (au)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{38}$
  - (av)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{39}$
  - (aw)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{40}$
  - (ax)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{41}$
  - (ay)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{42}$
  - (az)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{43}$
  - (ba)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{44}$
  - (bb)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{45}$
  - (bc)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{46}$
  - (bd)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{47}$
  - (be)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{48}$
  - (bf)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{49}$
  - (bg)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{50}$
  - (bh)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{51}$
  - (bi)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{52}$
  - (bj)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{53}$
  - (bk)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{54}$
  - (bl)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{55}$
  - (bm)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{56}$
  - (bn)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{57}$
  - (bo)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{58}$
  - (bp)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{59}$
  - (bq)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{60}$
  - (br)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{61}$
  - (bs)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{62}$
  - (bt)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{63}$
  - (bu)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{64}$
  - (bv)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{65}$
  - (bw)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{66}$
  - (bx)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{67}$
  - (by)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{68}$
  - (bz)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{69}$
  - (ca)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{70}$
  - (cb)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{71}$
  - (cc)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{72}$
  - (cd)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{73}$
  - (ce)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{74}$
  - (cf)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{75}$
  - (cg)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{76}$
  - (ch)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{77}$
  - (ci)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{78}$
  - (cj)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{79}$
  - (ck)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{80}$
  - (cl)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{81}$
  - (cm)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{82}$
  - (cn)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{83}$
  - (co)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{84}$
  - (cp)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{85}$
  - (cq)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{86}$
  - (cr)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{87}$
  - (cs)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{88}$
  - (ct)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{89}$
  - (cu)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{90}$
  - (cv)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{91}$
  - (cw)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{92}$
  - (cx)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{93}$
  - (cy)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{94}$
  - (cz)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{95}$
  - (da)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{96}$
  - (db)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{97}$
  - (dc)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{98}$
  - (dd)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{99}$
  - (de)  $\text{C}_2\text{H}_6\text{N}_2\text{O}_{100}$

2. Write the Lewis adducts formed by the following acid-base reactions:



a. Write the equation:



b. Write the structure of  $\text{BF}_3 \cdot \text{C}_2\text{H}_6$ .

c. Is addition of  $\text{BF}_3$  to  $\text{C}_2\text{H}_6$  an exothermic reaction? Is it a Lewis acid-base reaction?

d. Write a suitable Lewis acid-base adduct of  $\text{BF}_3$  with an alkene or alkyne.

e. Write a suitable Lewis acid-base adduct of  $\text{BF}_3$  with an amine or an alcohol.

# Chapter 9 Review Stoichiometry Section 1 Answers

**Mike Jess**



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