

# SALT ANALYSIS

1) For  $MgSO_4$  - magnesium Sulphate:-

**Aim:-** To analyze the given salt for an anion and one cation present in it.

**Material Required:-** Boiling Tube, Test Tubes, Test Tube Holder, Test Tube Stand, delivery tube, Corks, Filter paper, Test paper.

**Procedure:-**

S.No.	Test	Observation	Inference
1)	Colour	White	
2)	Odour		
3)	Flame test	Greenish white	
4)	Wet $H_2SO_4$ test (Salt + conc. $H_2SO_4$ )	No ppt. form.	
5)	Char. $H_2SO_4$ test (Salt + conc. $H_2SO_4$ )	No.	

2) For  $Fe^{2+}Cl^-$ , Ferrous chloride :-

**Aim:-** To analyze the given salt for an anion and one cation present in it.

**Material Required:-** Boiling Tube, Test Tube, Test Tube Holder, Test Tube Stand, delivery tube, Corks, Filter paper, Test paper.

**Procedure:-**

S.No.	Test	Observation	Inference
		Brown reddish	$Fe^{2+}$ may be present
		Colourless	$NO_3^-$ , $S^{2-}$ , $CO_3^{2-}$ are absent
			$Ca^{2+}$ , $S^{2-}$ , $SO_4^{2-}$ , $NO_3^-$ are absent
			$Cl^-$ may be present
			$Cl^-$ confirmed

Class 12th

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