

Physical Science Chapter 6 Test Review Sheet 2014-15

Assessments- This is the portion of the test which you, the student, will be permitted to retest on.

On the assessment, students will be required to:

1. Write out the names and formulas for a binary ionic compound.
2. Write out the names and formulas for a covalent (molecular) compound.

Students will know the following prefixes used for naming covalent/molecular compounds:

Mono-	1	Di-	2	Tri-	3
Tetra-	4	Penta-	5	Hexa-	6
Hepta-	7	Octa-	8	Nona-	9
Deca-	10				

Multiple Choice/Short Answer/Essay

1. Students will know, understand, and be able to apply the following terms:

electron dot diagram	ion	anion	cation
chemical bond	ionic bond	chemical formula	
covalent bond	molecule	polar covalent bond	metallic bond
polyatomic ion	semiconductor	molecular formula	empirical formula
hydrogen bond	formula unit	nonpolar covalent bond	

2. Students will know the composition, physical and chemical properties of ionic and covalent compounds.
3. Students will be able to predict and explain why elements lose or gain electrons to form ions.
4. Students will describe which groups in the periodic table are most willing to lose or gain electrons and explain why.
5. Students will know the different diatomic molecules as described in class (incomplete list in textbook)
6. Students will know what factors play a role in whether a molecular compound will be polar or nonpolar.

Physical Science Chapter 6 Test

AN Whitehead



Physical Science Chapter 6 Test:

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

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Climate Change 2007 - The Physical Science Basis Intergovernmental Panel on Climate Change, Intergovernmental Panel on Climate Change. Working Group I., 2007-09-10 The Climate Change 2007 volumes of the Fourth Assessment Report of the Intergovernmental Panel on Climate Change IPCC provide the most comprehensive and balanced assessment of climate change available This IPCC Working Group I report brings us completely up to date on the full range of scientific aspects of climate change Written by the world s leading experts the IPCC volumes will again prove to be invaluable for researchers students and policymakers and will form the standard reference works for policy decisions for government and industry worldwide Holt Physical Science Mapi M. Cuevas, 1994 *Motion, Forces, and Energy* Michael J. Padilla, Peter Kahan, 2002 *Teacher's Manual to Accompany Physical Science* Frederick J. Bueche, 1972 Physical science Jay M. Pasachoff, LeVon Balzer, Phyllis L. Goodson, Ann Collins, 1983 **Study Guide to Accompany: Fundamentals of Physical Science Six Edition** Arthur Beiser, 1971 *Focus on physical science* Charles H. Heimler, 1989

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