

Assessment 3

Chapter Test B**Chapter: Chemical Equations and Reactions**

Part 1 In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The production of a slightly soluble solid compound in a double-displacement reaction results in the formation of a
- gas.
 - precipitate.
 - redox reaction.
 - hydrogen.
- _____ 2. To balance a chemical equation, it is permissible to adjust the
- coefficients.
 - subscripts.
 - formula of the products.
 - number of products.
- _____ 3. In a chemical equation, the symbol (aq) indicates that the substance is
- solid.
 - dissolved in water.
 - an acid.
 - molten.
- _____ 4. The tendency for a replacement reaction to occur depends on the
- activity between any two elements in the activity series themselves.
 - temperature difference.
 - valence electrons and mass of.
 - activity between any two elements in the activity series themselves.
- _____ 5. The coefficients in a chemical equation
- indicate the number of moles of each substance that react.
 - show the number of grams of each substance that react.
 - are the molar masses of the substances.
 - show the valence electrons for each atom.
- _____ 6. If metal X is more reactive than metal Y in the activity series, then
- X will replace ions of Y in solution.
 - Y will replace ions of X in solution.
 - Y will form cations only indirectly.
 - X will react with cold water.

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AW Rasmussen



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