



COVALENT BONDING

Practice Problems

Do your best to solve each of the following problems.

SECTION 8.1 MOLECULAR COMPOUNDS

- Draw the Lewis structure for each of the following molecules.
 - H_2O
 - CO_2
 - H_2S
 - CH_4
 - NH_3
 - H_2O_2
 - C_2H_6
 - C_2H_4
 - C_2H_2
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 - H_2O_2
 - C_2H_6
 - C_2H_4
 - C_2H_2

SECTION 8.2 THE NATURE OF COVALENT BONDING

- Draw the Lewis structure for each of the following molecules.

$\text{H}-\text{F}$
- Draw the Lewis structure for each of the following molecules.

$\begin{array}{c} \text{H} \\ | \\ \text{H}-\text{C}-\text{H} \\ | \\ \text{H} \end{array}$
- Draw the Lewis structure for each of the following molecules.

$\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw the Lewis structure for each of the following molecules.

$\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw the Lewis structure for each of the following molecules.

$\text{H}-\text{C}\equiv\text{C}-\text{H}$



SECTION 8.3 BONDING: THE ORBITALS

- Draw the Lewis structure for each of the following molecules.
 - H_2O
 - CO_2
 - H_2S
 - CH_4
 - NH_3
 - H_2O_2
 - C_2H_6
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Chapter 8 Covalent Bonding Practice Problems Answers

Daniela Niemeyer



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