

Chapter Test B

Chapter: Chemical Equations and Reactions

PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The production of a slightly soluble solid compound in a double-displacement reaction results in the formation of a
- gas.
 - precipitate.
 - combustion reaction.
 - halogen.
- _____ 2. To balance a chemical equation, it is permissible to adjust the
- coefficients.
 - subscripts.
 - formulas of the products.
 - number of products.
- _____ 3. In a chemical equation, the symbol (aq) indicates that the substance is
- water.
 - dissolved in water.
 - an acid.
 - insoluble.
- _____ 4. The tendency for a replacement reaction to occur increases as the
- interval between any two elements in the activity series decreases.
 - temperature decreases.
 - valence electrons are used up.
 - interval between any two elements in the activity series increases.
- _____ 5. The coefficients in a chemical equation
- indicate the number of moles of each substance that react.
 - show the number of grams of each substance that react.
 - are the molar masses of the substances.
 - show the valence electrons for each atom.
- _____ 6. If metal X is lower than metal Y in the activity series, then
- X will replace ions of Y in solution.
 - Y will replace ions of X in solution.
 - Y will form oxides only indirectly.
 - X will react with cold water.

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Camilla Rothe



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