

Name \_\_\_\_\_

Molecular Geometries Lab Report Sheet

| Formula                      | Lewis Structure   | No of electron domains/Atp | Electron domain geometry     | VSEPR<br>Molecular geometry | Hybridization of central atom | Polar ? |
|------------------------------|---|----------------------------|------------------------------|-----------------------------|-------------------------------|---------|
| CO <sub>2</sub>              | <p>Valence 24</p> $\text{O}=\text{C}=\text{O}$ <p>2 charge clouds</p> | 0                          | Linear<br>180°               | Linear<br>                  | sp                            | NP      |
| BF <sub>3</sub>              | <p>Valence 32</p> $\text{F}=\text{B}-\text{F}$                        | 0                          | Trigonal planar<br>120°      |                             | sp <sup>2</sup>               | NP      |
| SnCl <sub>2</sub>            | <p>Valence 24</p> $\text{Cl}=\text{Sn}-\text{Cl}$                     | 3                          | Bent<br>120°                 |                             | sp <sup>2</sup>               | P       |
| BF <sub>4</sub> <sup>-</sup> | <p>Valence 40</p> $\left[ \text{F}-\text{B}-\text{F} \right]^{-}$     | 0                          | Tetrahedral<br>109.5°        |                             | sp <sup>3</sup>               | P       |
| AsH <sub>3</sub>             | <p>Valence 24</p> $\text{H}-\text{As}-\text{H}$                       | 4                          | Trigonal pyramidal<br>109.5° |                             | sp <sup>3</sup>               | P       |
| H <sub>2</sub> S             | <p>Valence 14</p> $\text{H}-\text{S}-\text{H}$                        | 2                          | Bent<br>109.5°               |                             | sp <sup>3</sup>               | NP      |
| NH <sub>2</sub> <sup>-</sup> | <p>Valence 10</p> $\left[ \text{H}-\text{N}-\text{H} \right]^{-}$     | 2                          | Bent<br>109.5°               |                             | sp <sup>3</sup>               | P       |
| PCl <sub>4</sub>             | <p>Valence 40</p> $\text{F}-\text{P}-\text{Cl}$                       | 0                          | Trigonal bipyre<br>120 + 90  |                             | sp <sup>3</sup> d             | P       |

# Lewis Structure And Molecular Models Lab Answers

**United States. Energy Research and  
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## **Lewis Structure And Molecular Models Lab Answers:**

## Embracing the Track of Appearance: An Emotional Symphony within **Lewis Structure And Molecular Models Lab Answers**

In a world taken by displays and the ceaseless chatter of instant interaction, the melodic splendor and emotional symphony developed by the prepared term usually disappear into the background, eclipsed by the persistent noise and distractions that permeate our lives. But, nestled within the pages of **Lewis Structure And Molecular Models Lab Answers** a wonderful fictional treasure brimming with fresh thoughts, lies an immersive symphony waiting to be embraced. Constructed by an elegant musician of language, this fascinating masterpiece conducts viewers on a mental journey, skillfully unraveling the concealed melodies and profound influence resonating within each carefully crafted phrase. Within the depths of this poignant evaluation, we shall explore the book's main harmonies, analyze its enthralling publishing fashion, and surrender ourselves to the profound resonance that echoes in the depths of readers' souls.

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