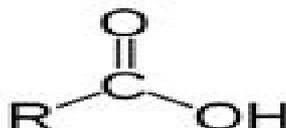


Please answer each question carefully. Remember, follow your gut! Please copy all answers on the answer sheet on the back of this exam! Otherwise, I will take off 10 points!

- The structure of DNA described by Watson and Crick included:
  - a double helix
  - the sugar-phosphate backbone aligned in the center of the helix
  - the base pairs that are stacked on the inside of the double helix
  - A double helix and base pairs that are stacked on the inside of the double helix
- What gives proteins such a dominant role in biochemistry?
  - the ability to act as a blueprint
  - their ability to spontaneously fold into complex three-dimensional structures
  - the variation in protein sizes
  - their ability to self-replicate
- Match the loss of a particular organelle with the associated disease.
  - Hypercholesterolemia - smooth endoplasmic reticulum
  - Diabetes - endosome
  - Tay-Sachs disease - lysosome
  - Muscle degeneration - mitochondria
- In a biochemistry lab course, you are asked to design an experiment to identify a strain of bacteria. Your lab partner claims that she thinks the bacterium contains a rough endoplasmic reticulum. To verify her claim, which of the following experiments would you perform? (pretend you're indulging her ludicrous claim).
  - determine whether the bacterium has an internal membrane-enclosed compartment
  - determine whether the bacterium generates  $\text{CO}_2$  in the presence of fuel molecules
  - determine whether the bacterium can synthesize proteins
  - determine whether the bacterium can synthesize ATP in the presence of fuel molecules and  $\text{O}_2$
- What is the  $\text{H}^+$  concentration in a urine sample that has a pH of 6?
  - $10^{-6} \text{ M}$
  - $10^{-8} \text{ M}$
  - $10^6 \text{ M}$
  - $10^{-14} \text{ M}$
- Which of the following is considered a noncovalent bond?
  - electrostatic interactions
  - hydrogen bonds
  - van der Waals interactions
  - All of these answers
- Which functional group is shown below?



- Amine
  - Thiol
  - Carboxylic acid
  - None of these answers
- What is the  $[\text{A}^-]/[\text{HA}]$  ratio when the weak acid is in a solution one pH unit above its  $\text{pK}_a$ ?

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### **Table of Contents Biochemistry Short Answer Questions**

1. Understanding the eBook Biochemistry Short Answer Questions
  - The Rise of Digital Reading Biochemistry Short Answer Questions
  - Advantages of eBooks Over Traditional Books
2. Identifying Biochemistry Short Answer Questions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Biochemistry Short Answer Questions
  - User-Friendly Interface
4. Exploring eBook Recommendations from Biochemistry Short Answer Questions

- Personalized Recommendations
  - Biochemistry Short Answer Questions User Reviews and Ratings
  - Biochemistry Short Answer Questions and Bestseller Lists
5. Accessing Biochemistry Short Answer Questions Free and Paid eBooks
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    - Biochemistry Short Answer Questions eBook Subscription Services
    - Biochemistry Short Answer Questions Budget-Friendly Options
  6. Navigating Biochemistry Short Answer Questions eBook Formats
    - ePub, PDF, MOBI, and More
    - Biochemistry Short Answer Questions Compatibility with Devices
    - Biochemistry Short Answer Questions Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Biochemistry Short Answer Questions
    - Highlighting and Note-Taking Biochemistry Short Answer Questions
    - Interactive Elements Biochemistry Short Answer Questions
  8. Staying Engaged with Biochemistry Short Answer Questions
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Biochemistry Short Answer Questions
  9. Balancing eBooks and Physical Books Biochemistry Short Answer Questions
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Biochemistry Short Answer Questions
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Biochemistry Short Answer Questions
    - Setting Reading Goals Biochemistry Short Answer Questions
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Biochemistry Short Answer Questions

- Fact-Checking eBook Content of Biochemistry Short Answer Questions
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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