

CONFORMATIONAL ANALYSIS: PRACTICE EXERCISES

10. Draw a Newman projection of the most stable conformation of 2-methylpropane.

11. The structures below are



- A) ethane.
- B) conformational isomers.
- C) enantiomers.
- D) structural isomers.
- E) both B and C.

12. Define the term *conformation*.

13. Draw a *staircase* molecule along the  $C_2-C_3$  bond and provide a *Newman* projection of the lowest energy conformation.

14. Provide a representation of the *gauche* conformation of butane.

15. Among the butane conformers, which occur at energy minima on a graph of potential energy versus dihedral angle?

- A) gauche only.
- B) eclipsed and totally eclipsed.
- C) anti and anti.
- D) eclipsed only.
- E) anti only.

16. Which of the following substituents exhibits the greatest negative heat of combustion per  $\text{CH}_2$  group?

- A) cyclohexane B) cyclopentane C) cyclohexene D) cyclobutane E) cyclopropane

17. Which of the following correctly ranks the substituents in order of increasing the heat per methylene ( $\text{CH}_2$ ) group?

- A) cyclohexane > cyclobutane > cyclohexene > cyclopropane
- B) cyclohexane > cyclopentane > cyclobutane > cyclopropane
- C) cyclohexane > cyclohexene > cyclobutane > cyclopropane
- D) cyclohexane > cyclopropane > cyclobutane > cyclohexene
- E) cyclopropane > cyclopentane > cyclobutane > cyclohexane

18. Describe the number of angles which will be formed when propan-2-ol is cyclopropaned.

# Conformational Analysis Practice Exercises

**N Colangelo**



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