

KINETICS Practice Problems and Solutions

Name: _____
 Period: _____ Date: _____

AP Chemistry
 Dr. Mandes

The following questions represent potential types of quiz questions. Please answer each question completely and thoroughly. The solutions will be posted on-line on Monday.

5. Please do #18 in chapter 12 of your text.
- This increases the $[H_2]$, which will increase the rate, but has no effect on k
 - Due to Arrhenius equation, changing temperature changes the value of k .
 - Catalyst function by lowering the activation energy, so due to Arrhenius equation, changing the activation energy changes the value of k .

6. Consider the reaction: $P_4 + 6 H_2 \rightarrow 4 PH_3$. A rate study of this reaction was conducted at 298 K. The data that were obtained are shown in the table.

$[P_4]$, mol/L	$[H_2]$, mol/L	Initial Rate, mol/L · s
0.0110	0.0075	3.20×10^{-4}
0.0110	0.0150	6.40×10^{-4}
0.0220	0.0150	6.39×10^{-4}

- What is the order with respect to: P_4 0
 H_2 1
- Write the rate law for this reaction. $rate = k[H_2]$
- Determine the value and units of the rate constant, k . **plug and chug using the rate law & data from exp't 1 and solving for k , we get $k = 0.0427 s^{-1}$**

7. Consider the reaction: $SO_2 + O_3 \rightarrow SO_3 + O_2$. A rate study of this reaction was conducted at 298 K. The data that were obtained are shown in the table.

$[SO_2]$, mol/L	$[O_3]$, mol/L	Initial Rate, mol/L · s
0.25	0.40	0.118
0.25	0.20	0.118
0.75	0.20	1.062

- What is the order with respect to: SO_2 2
 O_3 0
- Write the rate law for this reaction. $rate = k[SO_2]^2[O_3]^0$
- Determine the value and units of the rate constant, k . **plug and chug using the rate law & data from exp't 1 and solving for k , we get $k = 2.36 molL^{-2} s^{-1}$**

8. Consider the following mechanism.
- $$A_2 + B_2 \rightarrow R + C \quad (\text{slow})$$
- $$A_2 + R \rightarrow C \quad (\text{fast})$$

- Write the overall balanced chemical equation. $2 A_2 + B_2 \rightarrow 2 C$
- Identify any intermediates within the mechanism. **R**
- What is the order with respect to each reactant? **A_2 1st; B_2 1st**

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C Cleary



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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific, detailed, step-by-step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid-base reactions, which often persists through working with acid-base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields.

Technical Mechanics, Statics, Kinematics, Kinetics Edward Rose Maurer, Raymond Jefferson Roark, 1925
[Radiochemical Analysis of Agrochemicals](#) Mahbulul A. F. Jalal, Svetlana Bondarenko, 2025-03-06 This book provides an in-depth overview of agrochemicals, their development, regulation, and environmental impact. It also offers detailed insights into the latest research methodologies and regulatory standards. Through this comprehensive guide, readers will gain a thorough understanding of the complexities surrounding agrochemical use and its implications for modern agriculture. The chapters cover a wide range of topics, including the history and evolution of crop protection methods, the regulatory framework governing agrochemicals, and the intricate processes of degradation and metabolism in various environments. Expert contributors discuss the nuances of radioactivity in agrochemical studies, quantitative and spectroscopic analysis techniques, and the behaviour of these chemicals in soil, water, plants, and livestock. Given its breadth, this book appeals to academics, practitioners, policymakers, and students interested in understanding the scientific principles and practical applications of agrochemicals. Researchers in the fields of agricultural science, environmental chemistry, and regulatory affairs will find this book invaluable.

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Chemical Process Industries Louis Theodore,R. Ryan

Dupont,2022-12-30 The rapid growth and expansion of the chemical process industry during the past century have been accompanied by a simultaneous rise in human health problems as well as material and property losses because of fires explosions hazardous and toxic spills equipment failures other accidents and business interruptions Concern over the potential consequences of emissions of harmful chemicals along with catastrophic accidents has sparked interest at both the industrial and regulatory levels in obtaining a better understanding of the potential for environmental health risks in chemical and related industries This practical book presents and examines the environmental and health risk assessment calculations as they apply to various chemical process industries Chemical Process Industries Environmental and Health Risk Calculations can be used as a college text designed to provide new engineers and scientists some comprehension of the industries into which they may enter It also serves as a useful reference for practitioners and will help them better understand the health risk aspects of various industrial operations The chemical process industries employ mechanical electrical and civil engineers and a host of other scientists these professions should also benefit from material in this book that applies to their fields of work Kinetics and Catalysis ,1983-07 **Roadmap to the Regents** Sasha Alcott,2003 If Students Need to Know It It s in This Book This book develops the chemistry skills of high school students It builds skills that will help them succeed in school and on the New York Regents Exams Why The Princeton Review We have more than twenty years of experience helping students master the skills needed to excel on standardized tests Each year we help more than 2

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