

Calculations need to be based on the limiting reactant.

- Example 2: What is the maximum mass of sulfur dioxide that can be produced by the reaction of 95.6 g carbon disulfide with 100. g oxygen?



$$\frac{95.6 \text{ g } CS_2}{76.2 \text{ g } CS_2} \times \frac{1 \text{ mol } CS_2}{1 \text{ mol } CS_2} \times \frac{2 \text{ mol } SO_2}{1 \text{ mol } CS_2} \times \frac{64.1 \text{ g } SO_2}{1 \text{ mol } SO_2} = 161 \text{ g } SO_2$$

$$\frac{100. \text{ g } O_2}{32.0 \text{ g } O_2} \times \frac{1 \text{ mol } O_2}{3 \text{ mol } O_2} \times \frac{2 \text{ mol } SO_2}{3 \text{ mol } O_2} \times \frac{64.1 \text{ g } SO_2}{1 \text{ mol } SO_2} = 134 \text{ g } SO_2$$

Which reactant is **LIMITING**?

Limiting Reactant Problems And Solutions

D Keegan



Limiting Reactant Problems And Solutions:

Whispering the Secrets of Language: An Mental Journey through **Limiting Reactant Problems And Solutions**

In a digitally-driven world where monitors reign supreme and immediate connection drowns out the subtleties of language, the profound techniques and psychological subtleties hidden within words often move unheard. However, nestled within the pages of **Limiting Reactant Problems And Solutions** a interesting literary treasure pulsing with raw thoughts, lies a fantastic journey waiting to be undertaken. Composed by a skilled wordsmith, this marvelous opus invites readers on an introspective journey, softly unraveling the veiled truths and profound impact resonating within ab muscles fabric of each and every word. Within the psychological depths with this emotional review, we can embark upon a honest exploration of the book is core themes, dissect their captivating publishing fashion, and succumb to the effective resonance it evokes heavy within the recesses of readers hearts.

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