

Chemistry Chapter 8 Exam

Multiple Choice

1. Which of the following is not a Lewis acid?

- _____ A. CH_3COOH
- _____ B. Lewis Acids
- _____ C. Carbonyl Compounds
- _____ D. Alcohols
- _____ E. Lewis Bases
- _____ F. Simple Molecules
- _____ G. Multiple Bonds
- _____ H. Lewis Acids
- _____ I. Hydrogen Ion
- _____ J. None

- _____ A. bonded by coordinate covalent bonds that are superimposed on other bonds
- _____ B. single group of covalently bonded atoms
- _____ C. the energy released when one mole of an ionic crystalline compound dissolves in water
- _____ D. coordinate covalent bonds are part of structure in which both atoms share electrons
- _____ E. the energy required to break a chemical bond and form gaseous ions
- _____ F. structural group of atoms that are held together by covalent bonds, usually with the sharing of electron pairs between atoms
- _____ G. chemical bonding that results from the electrical attraction between positive and negative charges
- _____ H. mutual electrical attraction between protons and electrons, and between protons and negative atoms, including the atoms' negative charges

Multiple Choice

2. The charge on the ion is _____

- _____ A. always positive
- _____ B. always negative
- _____ C. either positive or negative
- _____ D. zero

3. In many compounds, atoms of one group form ions from bonds with the atoms of elements in the same group. This is _____

- _____ A. ionic bonding
- _____ B. covalent bonding
- _____ C. metallic bonding
- _____ D. none of these

4. Which of the following is not a Lewis acid? _____

- _____ A. CH_3COOH
- _____ B. Lewis Acids
- _____ C. Carbonyl Compounds
- _____ D. Alcohols
- _____ E. Lewis Bases
- _____ F. Simple Molecules
- _____ G. Multiple Bonds
- _____ H. Lewis Acids
- _____ I. Hydrogen Ion
- _____ J. None

Modern Chemistry Chapter 6 Test Answers

R Sandford



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