

Chapter 17: Chemical Equilibrium

Worksheet 17.1 (2017) 1-14

1. Consider the following reaction:

Reaction

- 1. The reaction is exothermic and the equilibrium constant is $K_c = 1.5$.
- 2. At equilibrium, the concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 3. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 4. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 5. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 6. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 7. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 8. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 9. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 10. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .

Answers

- 1. The reaction is exothermic and the equilibrium constant is $K_c = 1.5$.
- 2. At equilibrium, the concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 3. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 4. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 5. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
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- 8. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 9. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .
- 10. The concentration of CO_2 is 0.5 M and the concentration of CO is 1.0 M .

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