

**PERIODIC TABLE**  **STRAND: UNDERSTANDING PERIODICITY** **ACTIVITY**

**Section 8.1** *(continued)*

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the first row on the left below from the periodic table to complete the table on the right.


atomic mass	
atomic number	
electron configuration	
chemical name	
chemical symbol	

Put each item in Column A, under the letter of the matching item in Column B.

**Column A**

**Column B**

- |   |                            |
|---|----------------------------|
| _____ 18. A column in the periodic table                  | a. metals                  |
| _____ 19. A row in the periodic table                     | b. group                   |
| _____ 20. Group 1 elements                                | c. period                  |
| _____ 21. Elements that are shiny and conduct electricity | d. representative elements |
| _____ 22. Group 8 elements                                | e. transition elements     |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- \_\_\_\_\_ 23. There are two representative elements of platinum.
- \_\_\_\_\_ 24. Most elements besides boron of the elements in the periodic table are metalloids.
- \_\_\_\_\_ 25. Group 1A elements except for hydrogen are located on the left side.
- \_\_\_\_\_ 26. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 28. Group 8A elements are (a) nonmetals known as noble gases.
- \_\_\_\_\_ 29. Metalloids have properties of both metals and nonmetals.

# Chapter 6 Study Guide For Content Mastery Answers

**Freida L. Carson**



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