

**CHEMFILE MINI-GUIDE TO PROBLEM SOLVING**

**2. PLAN**

- *What step is needed to calculate the concentration of the diluted solution?*

Apply the principle that  $volume_1 \times molarity_1 = volume_2 \times molarity_2$ .

$$M_1V_1 = M_2V_2 \xrightarrow[\text{substitute each of the known quantities for its symbol and calculate}]{\begin{array}{l} \text{rearrange the equation} \\ M_1V_1 = M_2V_2 \\ \text{algebraically to solve for } M_2 \end{array}} M_2 = \frac{M_1V_1}{V_2}$$

$$M_2 = \frac{M_1 \times V_1}{V_2}$$

**3. COMPUTE**

Note: Even though molarity is moles per liter, you can use volumes in milliliters along with molarity whenever the units cancel.

$$M_2 = \frac{4.74 \text{ M} \times 50.00 \text{ mL}}{250.00 \text{ mL}} = 0.948 \text{ M}$$

**4. EVALUATE**

- *Are the units correct?*
- *Is the number of significant figures correct?*
- *Is the answer reasonable?*

Yes; molarity (mol/L) was required.  
 Yes; the number of significant figures is correct because the data had a minimum of three significant figures.  
 Yes; the computation is the same as 4.74/5, which is a little less than 1.

**PRACTICE**

1. Complete the table below by calculating the missing value in each row.

	<b>Molarity of stock solution</b>	<b>Volume of stock solution</b>	<b>Molarity of dilute solution</b>	<b>Volume of dilute solution</b>	
a.	0.500 M KBr	20.00 mL	7 M KBr	100.00 mL	ans: 0.100 M
b.	1.00 M LiOH	? mL	0.075 M LiOH	500.00 mL	ans: 38 mL
c.	7 M HI	5.00 mL	0.0493 M HI	100.00 mL	ans: 0.986 M
d.	12.0 M HCl	0.250 L	1.8 M HCl	? L	ans: 1.7 L
e.	7.44 M NH <sub>3</sub>	? mL	0.093 M NH <sub>3</sub>	4.00 L	ans: 50. mL

# Chemfile Mini Guide To Problem Solving Answer Key

**Anthony S. Fauci**



## **Chemfile Mini Guide To Problem Solving Answer Key:**

Holt Chemistry File ,1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket

**Children's Books in Print** R R Bowker Publishing,Bowker,1999-12 *Books in Print Supplement* ,2002 *Scientific and Technical Aerospace Reports* ,1977 **Holt ChemFile Lab Program** ,2005 *The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success (Volume One) (Preliminary E Rachel Turoscy,2017-12-31* [The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success \(Volume Two\) \(Preliminary E Rachel Turoscy,2017-12-31](#) [The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success \(Volume Two\) \(First Edition Rachel Turoscy,2018-08-09](#) The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 2 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses The book provides easy to understand step by step instructions for solving general chemistry problems The book begins with chapters dedicated to solutions kinetics and liquids solids and phase changes In subsequent chapters the text covers important topics like equilibrium concentrations strong and weak acids and bases the Common Ion Effect and reaction mechanisms It also covers the equilibrium between a solid and its respective ions in a solution as well as the second law of Thermodynamics The text also addresses Gibbs Free Energy equilibrium constants and electrolysis calculations Each chapter contains sample problems and practice problems to help further understanding of how math and chemistry go hand in hand The Answer Key is an excellent resource for any undergraduate course that deals with the basic concepts of general chemistry *Problem Solving Guide and Workbook for Introductory Chemistry by Steve Russo, Mike Silver* Sandra Yancy McGuire,Steve Russo, Mike Silver,2002 Provides over 175 worked examples and more than 500 practice problems and quiz questions to help students develop and practice their problem solving skills [The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success \(Volume One\) \(First Edition Rachel Turoscy,2018-08-09](#) The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 1 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses The book provides easy to understand step by step instructions for solving general chemistry problems The book begins with chapters dedicated to problem solving methodology and unit conversions In subsequent chapters the text covers important topics like ionic and covalent bonding chemical formula calculations solubility and reactions in aqueous solution gases the first law of Thermodynamics Quantum theory and electron configuration It also covers periodic trends the Lewis Dot Structures and bonding theories Each chapter contains sample

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