

# The Mole & Stoichiometry

## Across

3. What branch of science was the mole used

5. a unit of amount, the amount of a substance containing  $6.02 \times 10^{23}$  particles

8. What is the same number as the atomic number?

9. the molar mass of an element is the same as it's

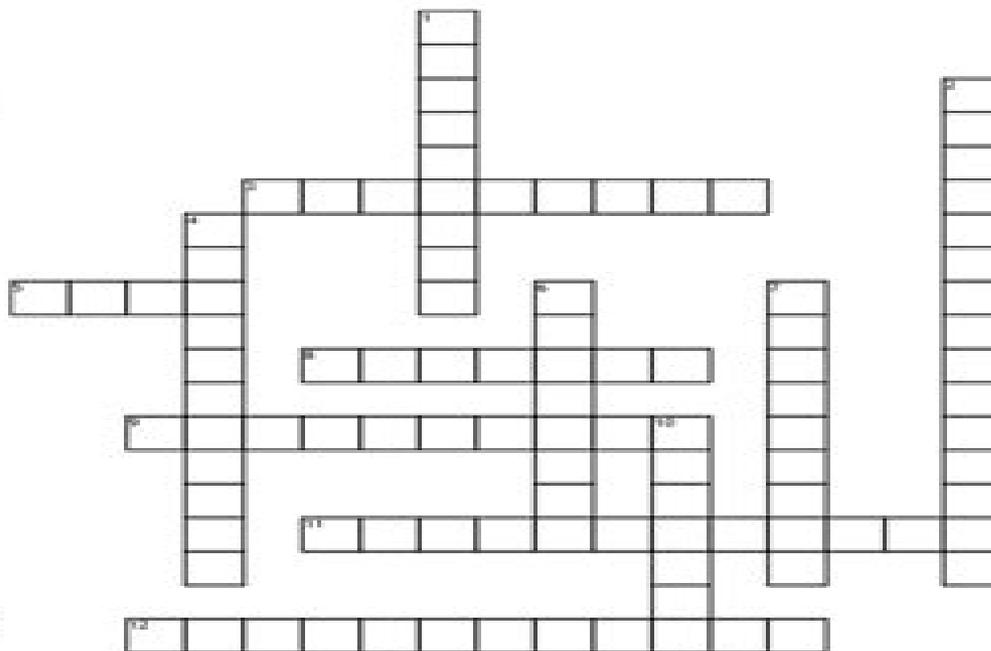
11. What serves as a standard for the mole unit

12. the numbers from a balanced chemical equation that determine mole ratios

## Down

1. The mass of one mole of a substance is called the

2. what do we call the number  $6 \times 10^{23}$



4. In Avogadro's number what is the 10 raised to?

6. who discovered the mole? (last name)

7. the moles measure the number of atoms, molecules or formula units all known as \_\_\_\_\_

10. What is the molar mass of Oxygen?



# Stoichiometry Crossword Answers

**RS Peters**



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