

Periodic Table Packet #1

Name _____

Period _____

Directions: Use a Periodic table to find the information asked for below:

1. What is the atomic number of:

Calcium _____
Iron _____
Gold _____
Uranium _____

2. What is the Atomic mass of:

Calcium _____
Iron _____
Uranium _____
Copper _____

3. How many protons do the following have?

Calcium _____
Gold _____
Copper _____
Iron _____

4. How many electrons do the following have?

Gold _____
Iron _____
Copper _____
Uranium _____

5. Does mercury have more protons and electrons than tin?

6. Is mercury a heavier element than tin?

7. Does potassium have more electrons than neon?

8. Does hydrogen have more electrons than Uranium?

9. Which has more protons, sulfur or iodine?

10. Write the symbols or the names for each of these elements:

Chlorine _____

Copper _____

Potassium _____

Silver _____

Na _____

Sn _____

Zn _____

Helium _____

Iron _____

P _____

Ne _____

Mercury _____

Periodic Table Packet 1 Answer Key

IM Harris



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Regents Chemistry--Physical Setting Power Pack Revised Edition Albert S. Tarendash,2021-01-05 Barron's two book Regents Chemistry Power Pack provides comprehensive review actual administered exams and practice questions to help students prepare for the Chemistry Regents exam This edition includes Regents Exams and Answers Chemistry Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day Let's Review Regents Chemistry Extensive review of all topics on the test Extra practice questions with answers A detailed introduction to the Regents Chemistry course and exam One actual recently released Regents Chemistry exam with an answer key

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