

- (i) Beta rays are made up of fast moving electrons.
- (ii) Velocity of beta particles is nearly equal to speed of light.
- (iii) Their ionizing power is less than that of Alpha particles.
- (iv) They have more penetrating power than Alpha particles.

PROPERTIES OF GAMMA RAYS

- (i) These rays are a type of electromagnetic waves.
- (ii) Rest mass of Gamma particles is zero.
- (iii) These rays have no charge.
- (iv) These rays are not affected by electric or magnetic field.
- (v) Speed of Gamma Rays is equal to speed of light.

13.20 | LAW OF RADIOACTIVITY

1st law: Radioactivity is a spontaneous process but it cannot be predicted that which atom will disintegrate first.

2nd law: In radioactive decay either α or β particles are emitted but both particle cannot emit simultaneously.

3rd law:

- During α decay mass no. reduced 4 and atomic number reduced by 2 during β decay.
- During β decay no change in mass no. by atomic no. reduce by 1
- During γ decay there is no change in mass no. as well as atomic no. but there will be change in energy state and during the nuclear transition the energy of γ rays is order of MeV.

4th law: (exponential decay law)

Acc. to this law the rate of disintegration of radioactive substance is directly proportional to no. of atoms un-decayed at the time.

$$\frac{dN}{dt} \propto N$$

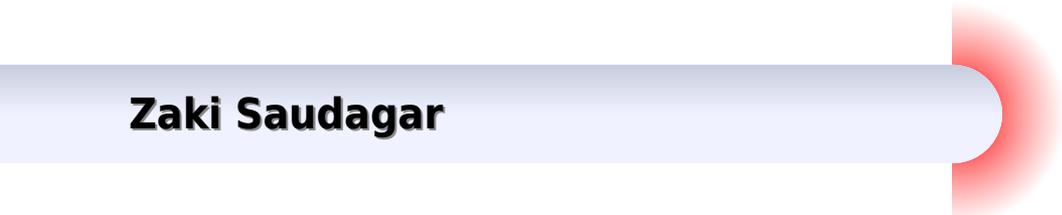
$$\frac{dN}{dt} = -\lambda N$$

$$\int_{N_0}^N \frac{dN}{N} = \int_0^t -\lambda dt$$

$$\log_e \frac{N}{N_0} = -\lambda t$$

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Table of Contents 12th Class Physics Notes Ecigarore

1. Understanding the eBook 12th Class Physics Notes Ecigarore
 - The Rise of Digital Reading 12th Class Physics Notes Ecigarore
 - Advantages of eBooks Over Traditional Books
2. Identifying 12th Class Physics Notes Ecigarore
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 12th Class Physics Notes Ecigarore
 - User-Friendly Interface
4. Exploring eBook Recommendations from 12th Class Physics Notes Ecigarore
 - Personalized Recommendations
 - 12th Class Physics Notes Ecigarore User Reviews and Ratings
 - 12th Class Physics Notes Ecigarore and Bestseller Lists

5. Accessing 12th Class Physics Notes Ecigarore Free and Paid eBooks
 - 12th Class Physics Notes Ecigarore Public Domain eBooks
 - 12th Class Physics Notes Ecigarore eBook Subscription Services
 - 12th Class Physics Notes Ecigarore Budget-Friendly Options
6. Navigating 12th Class Physics Notes Ecigarore eBook Formats
 - ePub, PDF, MOBI, and More
 - 12th Class Physics Notes Ecigarore Compatibility with Devices
 - 12th Class Physics Notes Ecigarore Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 12th Class Physics Notes Ecigarore
 - Highlighting and Note-Taking 12th Class Physics Notes Ecigarore
 - Interactive Elements 12th Class Physics Notes Ecigarore
8. Staying Engaged with 12th Class Physics Notes Ecigarore
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 12th Class Physics Notes Ecigarore
9. Balancing eBooks and Physical Books 12th Class Physics Notes Ecigarore
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 12th Class Physics Notes Ecigarore
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 12th Class Physics Notes Ecigarore
 - Setting Reading Goals 12th Class Physics Notes Ecigarore
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 12th Class Physics Notes Ecigarore
 - Fact-Checking eBook Content of 12th Class Physics Notes Ecigarore
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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