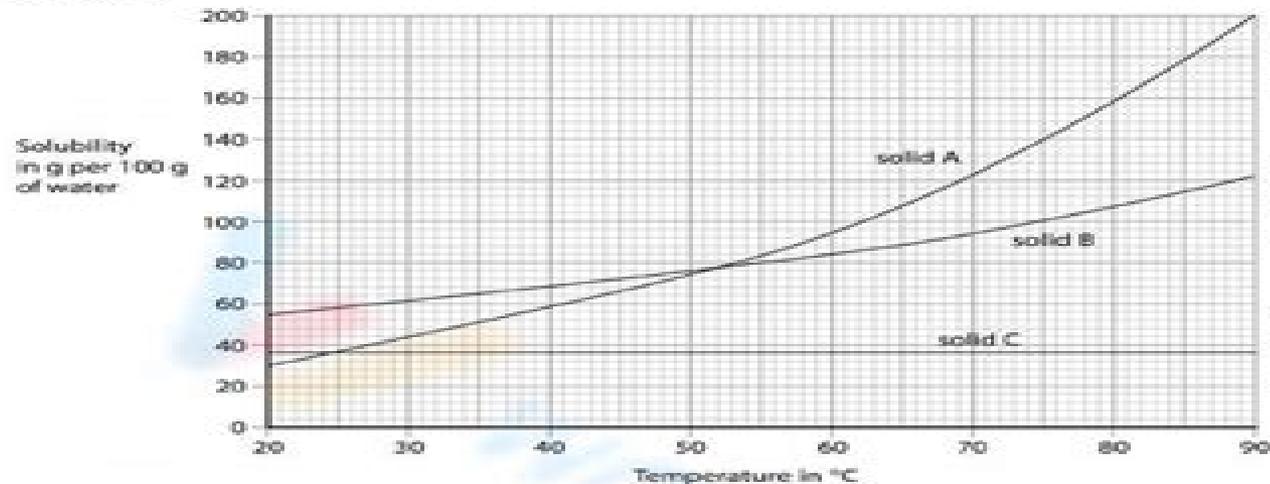


Use the Solubility Curve Graph to Answer the following. It is the same as the one in your notebook.



1. What is the Solubility of Solid A at 40°C?

2. What is the Solubility of Solid B at 30°C?

3. What is the Solubility of Solid C at 60°C?

4. How much of Solid A will dissolve in 100g of water at 80°C?

5. How much of Solid B will dissolve in 100g of water at 40°C?

6. Which Solid is the LEAST soluble at 60°C?

7. Which Solid is MOST soluble at 60°C?

8. At what temperature is the solubility of Solid A and Solid B the same?

9. Saturated - Unsaturated - Supersaturated: Solution that contains 70g of Solid B at 30°C?

10. Saturated - Unsaturated - Supersaturated: Solution that contains 110g of Solid A at 70°C?

11. How does temperature affect the solubility of solutions?

- As temperature increases, solubility decrease
- As temperature increases, solubility increases
- As temperature decreases, solubility increases

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M Mosston



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