



REACTING MASS CALCULATIONS - INTRODUCTION

- Step 1** Write ✓ for the substance whose mass is given and ? for the substance whose mass is to be calculated on the balanced equation
- Step 2** Find the moles of the ✓ substance (using $\text{moles} = \frac{\text{mass}}{M_r}$)
- Step 3** Use the balanced equation and your answer from step 2 to find the moles of the ? substance
- Step 4** Find the mass of the ? substance (using $\text{mass} = M_r \times \text{moles}$)

1) What mass of oxygen reacts with 12 g of magnesium?



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2) What mass of calcium hydroxide is made from 14 kg of calcium oxide?



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3) What mass of aluminium is needed to react with 640 g of iron oxide?



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4) What mass of titanium chloride reacts with 460 g of sodium?



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Calculations In Chemistry An Introduction

Peter W. Atkins, Ronald S. Friedman



Calculations In Chemistry An Introduction:

Calculations In Chemistry An Introduction Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Calculations In Chemistry An Introduction**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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