

PHYS2080 Equation Sheet

Thermodynamics

$$\sigma = 5.67 \times 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$$

$$k_B = 1.38 \times 10^{-23} \text{ J/K}$$

$$R = 8.31 \text{ J/(mol.K)}$$

$$T_K = T_C + 273 \text{ K,}$$

$$\Delta T_K = \Delta T_C$$

$$T_F = \frac{9}{5} T_C + 32^\circ \text{F,}$$

$$T_C = \frac{5}{9} (T_F - 32)$$

$$\Delta L = \alpha L_0 \Delta T$$

$$\Delta A \approx 2\alpha A_0 \Delta T$$

$$\Delta V = \beta V_0 \Delta T \approx 3\alpha V_0 \Delta T$$

$$P_{\text{conduction}} = \frac{Q}{\Delta t} = k \frac{A}{l} (T_{\text{hot}} - T_{\text{cold}})$$

$$P_{\text{radiated}} = \frac{Q}{\Delta t} = \sigma e A T^4$$

$$PV = nRT,$$

$$nR = Nk_B$$

$$n = \frac{m}{M} = N/N_A$$

$$Q = mc\Delta T$$

$$Q = \pm mL$$

$$\text{At constant pressure: } Q = nC_P \Delta T$$

$$\text{At constant volume: } Q = nC_V \Delta T$$

$$C_P = C_V + R$$

$$K_{\text{avg,translational}} = \frac{3}{2} k_B T$$

$$K_{\text{tot}} = \frac{3}{2} Nk_B T = U \text{ (monoatomic gas)}$$

$$v_{\text{rms}} = \sqrt{\frac{3RT}{M}} = \sqrt{\frac{3k_B T}{m}}$$

$$v_{\text{avg}} = \sqrt{\frac{8RT}{\pi M}}$$

$$v_{\text{mp}} = \sqrt{\frac{2RT}{M}}$$

$$\Delta U = Q - W$$

$$\text{Isobaric process: } W = P\Delta V$$

$$\text{Isochoric process: } W = 0$$

$$\text{Isothermal process: } W = nRT \ln \frac{V_f}{V_i} = Nk_B T \ln \frac{V_f}{V_i} = P_i V_i \ln \frac{V_f}{V_i} = P_f V_f \ln \frac{V_f}{V_i}$$

$$PV = \text{constant}$$

$$PV^\gamma = \text{constant}$$

$$\gamma = C_P / C_V$$

$$W = Q_h - Q_c$$

$$e = W / Q_h = 1 - \frac{|Q_c|}{Q_h}$$

$$e_{\text{Carnot}} = 1 - \frac{T_C}{T_H}$$

$$\text{COP: } K_{\text{Cooling}} = \frac{Q_c}{W_{\text{motor}}} = \frac{Q_c}{|Q_h| - Q_c}$$

$$K_{\text{Heating}} = \frac{|Q_h|}{W_{\text{motor}}} = \frac{|Q_h|}{|Q_h| - Q_c}$$

$$\text{Carnot COP } K_{\text{Carnot,Heating}} = \frac{T_h}{T_h - T_c}$$

$$K_{\text{Carnot,Cooling}} = \frac{T_c}{T_h - T_c}$$

Basic Geometry

$$A_{\text{sphere}} = 4\pi R^2$$

$$A_{\text{cube}} = 6a^2$$

$$A_{\text{triangle}} = \frac{1}{2} bh$$

$$V_{\text{cylinder}} = \pi R^2 h$$

$$V_{\text{sphere}} = \frac{4}{3} \pi R^3$$

$$V_{\text{cube}} = a^3$$

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Lingjun Ying



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Notes on Thermodynamics John Alonzo Goff, 1950 Basic Thermodynamics 3rd Semester Prof. Hurmathulla Khan, Dr. Manjunath Gowda M.R, Dr. Kiran Gowda M R, Prof. Arjun M, Prof. Hanumantharayappa D, Dr. Ashok R Banagar, 2025-07-01 Thermodynamics the science of energy transformations and the physical laws governing them plays a foundational role in understanding the natural world and engineering systems alike Whether it is the inner workings of an engine the cooling of electronic devices or the behavior of atmospheric systems thermodynamics offers powerful tools and principles that are as fundamental as they are far reaching This book Basic Thermodynamics is designed to introduce readers to the core concepts of classical thermodynamics in a clear systematic and accessible manner It is intended primarily for undergraduate students in mechanical chemical and aerospace engineering but also caters to curious learners and practitioners who seek a solid understanding of thermal sciences The chapters are arranged to build logically from basic definitions to more complex applications Beginning with an introduction to thermodynamic systems and properties the text moves through the first and second laws of thermodynamics entropy thermodynamic cycles and an overview of real world applications Emphasis has been placed on conceptual clarity supported by illustrative examples problem solving strategies and practical case studies Special care has been taken to explain the physical significance behind the mathematical formulations allowing students to connect equations with real world phenomena Each chapter ends with a set of exercises that reinforce key ideas and encourage deeper engagement with the material This book would not have been possible without the guidance and support of mentors colleagues and students whose insights and feedback have helped shape its content and structure I hope this work serves as a valuable stepping stone for readers on their journey through the fascinating and essential world of thermodynamics

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