

ELECTRON PROBABILITY LAB

The position of an electron in an atom at a given moment cannot be predicted. The region of space in which the electron can *probably* be found is called an **orbital**. Orbitals are often referred to as "**electron clouds**" because they are not absolute. In fact, there are times when an electron may be found outside the orbital.

The ground state orbital of hydrogen is represented by a fuzzy sphere with the nucleus at the center. The denser areas of sphere represent areas where there is a greater **probability** of finding an electron. Since the electron in a hydrogen atom is in constant motion, it may be helpful to compare the electron cloud to spinning fan blades. Even though there is a finite number of blades, they appear to fill the complete circle through which they turn. In the same manner, the single electron of a hydrogen atom appears to fill the entire electron cloud.

In this lab, you will use a felt-tip marker and a target to investigate the probability distribution of marks about a central point. This two-dimensional model will help you better understand the three-dimensional distribution of the electron in the ground state orbital of hydrogen.

PROCEDURE

1. Place your target on a notebook on the floor. The target should be flat. The notebook simply serves as a cushion.
2. Drop your marker from a height of about 1 meter onto the target so that it makes a mark. Aim for the center. Repeat this at least 100 times. (TIP: Do not throw the marker. Let the marker fall from a position that appears to be directly over the center of the target.)
3. Count the number of marks in each numbered region of the target and record the numbers in the data table. (TIP: Circle the marks with a pencil as you count them to avoid counting them twice.)
4. Return your marker. Clean any stray marks off the floor with a wet paper towel.

DATA

1. Record your mark counts in your lab report using the following data table.

REGION	NUMBER OF MARKS
0	0
1	
2	
3	
4	
5	
6	



2. The team secretary should attach the target in his/her lab book as shown in the diagram above. Use tape or glue. Other team members need to state whose book contains the target.

Electron Probability Lab Answers

Xiang Xie



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Who's the New Kid in Chemistry? John D. Butler, 2013-12-12 *Who's the New Kid in Chemistry* offers an unprecedented look at student engagement and teacher best practices through the eyes of an educational researcher enrolled as a public high school student Over the course of seventy nine consecutive days John D Butler participates in and observes Rhode Island 2013 Teacher of the Year Jessica M Waters's high school chemistry class documenting his experiences as they unfold *Who's the New Kid in Chemistry* is a compelling example of what can be accomplished when an educational researcher and teacher collaborate in the classroom This work includes a discussion on flexible homework assignments data driven instruction and thirty teacher best practices This book is an invaluable resource for teachers across all content areas masters and doctoral research method classes and future Teachers of the Year

Nuclear Science Abstracts, 1966 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 *Prentice Hall Physical Science Concepts in Action* helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

Prentice Hall Chemistry, 2000 *Government Reports Announcements & Index*, 1985 Quantities, Units, and Measuring Methods of Ionizing Radiation Franco Fossati, Lia Forti, 1959 **International Aerospace Abstracts**, 1974 *Government Reports Announcements*, 1971-11-25 **Proceedings**, 1964 *Japanese Current Research*, 1987-07 Merrill Chemistry Robert C. Smoot, Richard G. Smith, Jack Price, 1998 **Energy Research Abstracts**, 1993 Includes all works deriving from DOE other related government sponsored information and foreign nonnuclear information

Computer Mathematics, Series II Geoffrey Knight, 1969 General numerical and symbolic analysis Elementary algebra Calculus Difference differential and integral equations Abstracts mathematics Probability and statistics Optimization mathematical programming operations research Mathematical communication theory information theory Mathematical systems and control theory Mathematical logic and switching theory automata

Physics Briefs, 1994 Who's who in Technology Today Barbara A. Tinucci, 1984 **Annual Book of ASTM Standards** American Society for Testing and Materials, 1992 The Education Index, 1990 **INIS Atomindex**, 1987 Acronyms Dictionary Gale Research Company, 1960 **Radio Engineering & Electronic Physics**, 1968

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