

# GAS LAWS

Answer:  
Gas molecules are constantly in motion and hit other gas molecules and the walls of the container. For each of them, whether the number of collisions per second increases and then explain your answer.

1.  Double the number of molecules.

Collisions per second will increase/decrease

2.  Decrease the volume of the container.

Collisions per second will increase/decrease

3.  Increase the temperature.

Collisions per second will increase

Answer:  
Answer the questions below. Use the space under each set of questions for your work.

Class: \_\_\_\_\_ Date: \_\_\_\_\_

## Charles' Law

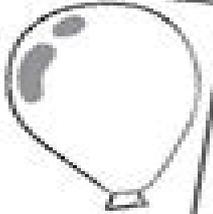
A fun-loving chemist has a balloon at standard temperature and pressure and the balloon has a volume of 3.00 L.



1. What is the initial volume of the balloon?
2. What is the existing temperature?

If the pressure is held constant and the balloon is heated to 400°C, what will the new volume of the balloon be?

3. What is the new temperature of the system?



4. What is the new volume of the balloon?

From the fun-loving chemist - placed it in a box of 4°C. If the...

# 18 CHEMISTRY HOMEWORK PAGES

# Unit 10 Gas Laws Homework Chemistry Answers

**Mike Stieff**



**Unit 10 Gas Laws Homework Chemistry Answers:**

## Whispering the Strategies of Language: An Emotional Quest through **Unit 10 Gas Laws Homework Chemistry Answers**

In a digitally-driven world where speed reigns supreme and quick conversation drowns out the subtleties of language, the profound strategies and mental nuances hidden within phrases frequently go unheard. Yet, set within the pages of **Unit 10 Gas Laws Homework Chemistry Answers** a charming fictional value sporting with organic feelings, lies an extraordinary journey waiting to be undertaken. Published by a talented wordsmith, this wonderful opus invites viewers on an introspective journey, delicately unraveling the veiled truths and profound impact resonating within the fabric of every word. Within the mental depths of this moving review, we will embark upon a heartfelt exploration of the book's core styles, dissect its fascinating publishing style, and fail to the powerful resonance it evokes deep within the recesses of readers' hearts.

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