

## Chemistry I (H) - Unit 02 (Pt. II - Nuclear Chemistry) Study Guide

- 1) Define, compare, and contrast radiation (pg. 122) to radioactivity (pg. 122).
- 2) Define AND give an example of a radioisotope (pg. 861).
- 3) Define nuclear reactions.
  - Define nuclear equations (pg. 123/869) -
  - Example:
- 4) Define and provide examples of the following forms of radioactive decay:
  - alpha decay (pg. 867) -
    - alpha particle (pg. 123/862) -
    - Example Nuclear Equation:
  - beta decay (pg. 867) -
    - beta particle (pg. 123/863) -
    - Example Nuclear Equation:
  - gamma radiation (pg. 124) -
    - gamma ray (pg. 124/863) -
    - Example Nuclear Equation:
  - Study the Table 24.2 from pg. 861 that we drew in our notes about the properties of decay (penetrating power, mass, etc.)
- 5) Define AND give an example of transmutation (pg. 865).
- 6) What do we mean by the words emit or emanate? Give an example.
- 7) Compare a *stable nucleus* to that of an *unstable nucleus*.

# Unit 17 Nuclear Chemistry Study Guide Answers

**Norman E. Griswold, Henry Fuller  
Holtzclaw**



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## **Unit 17 Nuclear Chemistry Study Guide Answers** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Unit 17 Nuclear Chemistry Study Guide Answers**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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