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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{\ln(N_0/N)}{\ln 2}$ $N = N_0 \times (0.5)^n$ 125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore $n = \frac{\ln(80/100)}{\ln 0.5}$ now using the half life equation in example 1 we have

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