

REPORT SHEET
Acid-Base Titration

LAB
10

A. Concentration of Acetic Acid in Vinegar

1. Brand Heinz Volume 5.00 mL (% on label) 5.0 %
 2. Molarity (M) of NaOH _____ M

| | Trial 1 | Trial 2 | Trial 3 |
|---------------------------------|---------|---------|---------|
| 3. Initial NaOH level in buret | 0.1 mL | 0.2 mL | 0.1 mL |
| 4. Final NaOH level in buret | 39.6 mL | 41.2 mL | 40.1 mL |
| 5. Volume (mL) of NaOH used | | | |
| 6. Average volume (mL) | | | |
| 7. Average volume in liters (L) | | | |

Show all calculations for the following questions.

8. Moles of NaOH used in titration _____ mole NaOH
9. Moles of $\text{HC}_2\text{H}_3\text{O}_2$ neutralized by NaOH _____ mole $\text{HC}_2\text{H}_3\text{O}_2$
10. Molarity of $\text{HC}_2\text{H}_3\text{O}_2$ _____ M $\text{HC}_2\text{H}_3\text{O}_2$
11. Grams of $\text{HC}_2\text{H}_3\text{O}_2$ _____ g $\text{HC}_2\text{H}_3\text{O}_2$
12. Percent (m/v) $\text{HC}_2\text{H}_3\text{O}_2$ in vinegar _____ % $\text{HC}_2\text{H}_3\text{O}_2$

Acid Base Titration Curve Lab Answers

**Thirumalaisamy P.
Velavancorresponding**



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