

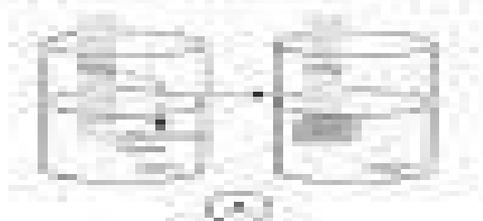
The Activity Series

Write a net ionic equation for each of the reactions indicated.

10 level

Write a net ionic equation for each of these reactions. Write a net ionic equation for each reaction. If no reaction occurs, write "no reaction" or "no equation." Remember, when writing a net ionic equation, all species already in ionic form are shown as ions, and all species that are formed are shown as ions. Write a net ionic equation for each reaction. If no reaction occurs, write "no reaction" or "no equation." If no reaction occurs, write "no reaction" or "no equation."

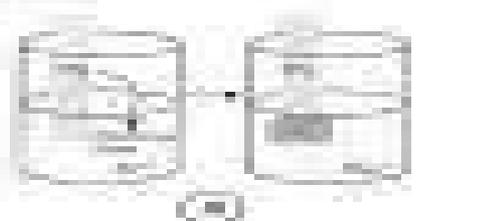
Model 1 – Reactivity of Metals



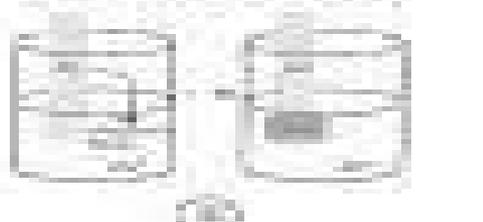
$\text{Zn} + \text{Ag}^+ \rightarrow \text{Zn}^{2+} + \text{Ag}$



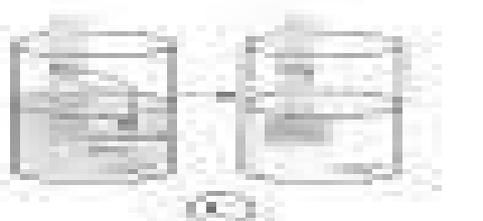
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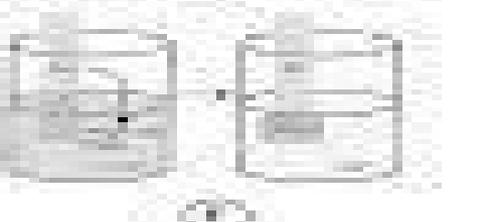
$\text{Cu} + \text{Ag}^+ \rightarrow \text{Cu}^{2+} + \text{Ag}$



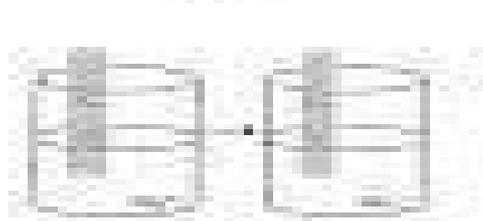
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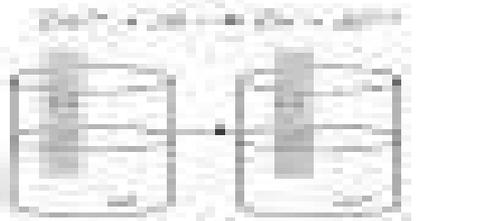
$\text{Ag} + \text{Ag}^+ \rightarrow \text{Ag} + \text{Ag}^+$



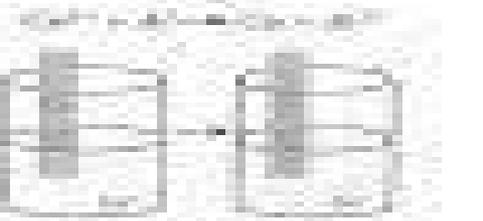
no



no



no



no

Note: These reactions and equations are simplified to focus on the displacement of electrons.

Activity Series Pogil Answers

G Thomas



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