

## Post-Lab Questions

1. You observed the relative reactivity of the following metals.
- Hg reacts with HNO<sub>3</sub> but not with HCl, hot or cold water.
  - Al reacts slowly with hot (100 °C) water.
  - Sn reacts with cold water.
  - Cd reacts with HCl but not with hot or cold water.

Write the mini-activity series (left to right) for these 4 metals (most active on right).

least active

most active

2. Indicate the products that would be produced (if any) when the following substances are mixed.



3. a. Write the symbols for the alkali metals in order of decreasing activity

\_\_\_\_\_ > \_\_\_\_\_ > \_\_\_\_\_ > \_\_\_\_\_ > \_\_\_\_\_

- b. Write the symbols for the alkaline earth metals in order of decreasing activity

\_\_\_\_\_ > \_\_\_\_\_ > \_\_\_\_\_ > \_\_\_\_\_ > \_\_\_\_\_

- c. Write the symbols for the Group 3a metals in order of increasing activity

\_\_\_\_\_ < \_\_\_\_\_ < \_\_\_\_\_ < \_\_\_\_\_ < \_\_\_\_\_

# Activity Series Post Lab Answers Experiment 7

**RJ Alexander**



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