JGi

**JAIN COLLEGE, J C Road Bangalore**
**Mock Paper -1, February - 2015**
Time: 3 Hours 15 Minutes     II PUC– Chemistry (34)     Max. Marks: 70

Instructions:
1. The question paper has four parts: A, B, C and D. All parts are compulsory.
2. Write balanced chemical equations and draw labelled diagrams wherever required.
3. Use log tables and the simple calculator if necessary. (Use of scientific calculators is not allowed).

## PART - A

I     **Answer ALL the following questions.**                                   10 X 1 = 10
      (Answer each question in one word or in one sentence)

1. Two liquids A and B on mixing produce a warm solution. Which type of deviation from Raoults law does it show?
2. What freezes out first when a solution of common salt is cooled?
3. Why SHE is named as a reversible electrode?
4. Write the rate equation for the reaction $2A + B \longrightarrow C$, if the order of the reaction is zero.
5. Sky appears blue in colour. Give reason.
6. Name the method used for the purification of Nickel.
7. Give one use of argon.
8. Give the IUPAC name of $CH_3CH=CHC(Br)(CH_3)_2$.
9. $2CH_3\text{-}COCl + (CH_3)_2Cd \longrightarrow X + CdCl_2$. What is X?
10. Write the structure of sugar present in RNA.

## PART - B

II     **Answer any FIVE of the following (Each question carries 2 marks).**     5 X 2 = 10
11. Account for the following:
    (a) Metallic crystals show high electrical conductivity.
    (b) Molecular crystals are soft and volatile
12. State Faradays second law.
13. Rate constant of a first order reaction is $2.3 \times 10^{-4} s^{-1}$. Calculate the time taken for the completion of $2/3^{rd}$ of the reaction.
14. Atomic sizes of 3d series elements from chromium to copper are almost same. Give reason.
15. How is salicylic acid converted to aspirin?
16. Describe haloform reaction with an example.
17. What are tranquilizers? Give examples.
18. What are artificial sweeteners? Which artificial sweetener is stable at cooking temperature?

## PART – C

III.     **Answer any FIVE of the following (Each question carries 3 marks).**     5 X 3 = 15
19. Explain the extraction of aluminium from purified alumina by Hall-Heroult process.
20. (a) How does copper react with dil. $HNO_3$?
    (b) Which allotropic form of phosphorous exhibits chemiluminescence?     (2+1)
21. Explain the steps involved in the manufacture of sulphuric acid by contact process.
22. (a) Complete the following reactions:
    (i)     $NH_3 + Cl_2 \longrightarrow$ _____ + _____
    (ii)    $Na_2SO_3 + Cl_2 + H_2O \longrightarrow$ _____ + _____
    (c) Hydrogen fluoride is liquid, but other hydrogen halides are gases. Give reason.     (2+1)

-1-

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