

5-2 Review and Reinforcement Part II

Quantum Theory and the Atom

Explain this principle: state the evidence for the discreteness of light (quantization).

photon(s)

Planck constant (symbol)

quantum mechanical model

quantum number(s)

electron density

orbitals

principal energy levels

_____ orbital _____ 1-4: model to explain around the nucleus where electrons in atoms really found.

_____ electron density _____ 2: the density of an electron cloud

_____ quantum number _____ 3: number describing a probability density found in atoms

Explain this principle: is there any fractional quantum number (orbitals, electrons, density) and why/why not?

_____ quantum number (orbitals) _____ 1: quantum number represents spin

_____ quantum number (orbitals) _____ 2: probability probability obtained by knowing the electron density and

_____ probability (orbitals)

_____ probability (orbitals) _____ 3: the total energy level in atoms

_____ electron spin _____ 4: the direction of counter-clockwise or clockwise rotation

Explain this principle: state the laws of the quantum mechanical model (electrons, density)

_____ 1. _____ 1: the electron cloud is everywhere where the probability of finding an electron is

- greater
- smaller
- higher density
- constant

_____ 2. _____ 2: the principal energy level of the hydrogen atom contains only one

- orbital
- or orbitals
- of orbitals
- orbitals

4 2 Review And Reinforcement Quantum Theory Answers

Minjie Lin



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