

## Scaffold



# Brand Services Inc Technical Manual Scaffolding

**H Kauffman**



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web equilibrium moles  $\text{HSCN}$   $2.00 \times 10^{-5}$  mol  $3.00 \times 10^{-6}$  mol  $1.70 \times 10^{-5}$  mol  $\text{HSCN}$  knowing the number of moles of  $\text{Fe}^{3+}$  and  $\text{HSCN}$  present in the equilibrium mixture and the volume of the mixture we can easily find the concentrations of those two species mol  $\text{Fe}^{3+}$   $3.170 \times 10^{-5}$  mol  $\text{Fe}^{3+}$   $3.850 \times 10^{-4}$  M

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web nov 13 2022 define the equilibrium state of a chemical reaction system what is its practical significance state the meaning and significance of the following terms reversible reaction quantitative reaction kinetically inhibited reaction explain the meaning of the statement equilibrium is macroscopically static but microscopically dynamic very

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web in a chemical equilibrium the forward and reverse reactions occur at equal rates and the concentrations of products and reactants remain constant if we run a reaction in a closed system so that the products cannot escape we often find the reaction does not give a 100 yield of products