

floats when the pressure is 500 kPa. This stop is now cooled to 100 °C by rejecting heat to the surrounding at 20 °C. Find the total entropy generated in the process.

System = Water

State 1 $\left. \begin{array}{l} P_1 = 1 \text{ MPa} \\ T_1 = 500^\circ\text{C} \end{array} \right\} m = \frac{V}{\rho}$

$\rho_{H_2O} = \checkmark$
 $V_1 = 3 \text{ m}^3$

final state, $T = 100^\circ\text{C}$

When pressure inside the system is red. to 500 kPa
 \hookrightarrow floats
 \hookrightarrow it starts to move.

$1 \text{ MPa} \rightarrow 500 \text{ kPa}$
 $V = \text{const.}$



2. When the system H.T. \hookrightarrow to 100 °C

So, At this condition piston will hit the lower

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