

12

STOICHIOMETRY

Practice Problems

In your notebook, solve the following problems.

SECTION 12.1 THE ARITHMETIC OF EQUATIONS

Use the 3-step problem-solving approach you learned in Chapter 1.

1. An apple pie needs 10 large apples, 2 cups rasp and blueberries, and 1 tablespoon of cinnamon. Write a balanced equation that fits this situation. How many apples are needed to make 20 pies?
2. Two moles of potassium chloride and three moles of oxygen are produced from the decomposition of two moles of potassium chlorate, KClO_3 . Write the balanced equation. How many moles of oxygen are produced from 12 moles of potassium chlorate?
3. Using the equation from problem 2, how many moles of oxygen are produced from 14 moles of potassium chlorate?
4. Two molecules of hydrogen react with one molecule of oxygen to produce two molecules of water. How many molecules of water are produced from 3.8×10^{22} molecules of oxygen? How many moles of water are produced from 22.5 moles of oxygen?

SECTION 12.2 CHEMICAL CALCULATIONS

1. Calculate the number of moles of hydrogen chloride produced from 10 moles of hydrogen.



2. Calculate the number of moles of chlorine needed to form 14 moles of iron(III) chloride.



3. Calculate the number of grams of nitrogen dioxide that are produced from 4 moles of nitric oxide.



4. Calculate the mass of oxygen produced from the decomposition of 15.0 g of potassium chlorate.



5. Calculate the mass of silver needed to react with chlorine to produce 94 g of silver chloride. (Write a balanced equation first.)

6. How many liters of carbon monoxide at STP are needed to react with 4.00 g of oxygen gas to produce carbon dioxide?



7. Calculate the number of liters of oxygen gas needed to produce 15.0 liters of nitrogen dioxide. Assume all gases are at the same conditions of temperature and pressure.



Pearson Chemistry Chapter 12 Stoichiometry

David Hendricks



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Applied Chemistry Siddharth Venkatesh, 2025-02-20 Discover the essential aspects of chemistry in various industries with Applied Chemistry Practical Applications This comprehensive textbook provides an in depth understanding of fundamental chemical principles and their real world applications Covering a wide range of topics from chemical reactions and materials science to environmental chemistry and sustainable practices it caters to students researchers and professionals Written by experts our book blends theoretical concepts with practical examples offering a solid foundation in key concepts followed by discussions on their applications in industry technology and everyday life We emphasize sustainability green chemistry principles and environmentally friendly practices Clear explanations of complex topics are supported by diagrams illustrations and tables Our book integrates modern research findings and technological advancements in chemistry End of chapter summaries review questions and exercises reinforce learning and facilitate self assessment Supplementary materials including online resources and laboratory exercises enhance the learning experience Whether you re a student seeking an introduction to applied chemistry or a professional looking to expand your knowledge Applied Chemistry Practical Applications is an invaluable resource for understanding the practical aspects of chemistry in industry technology and society

Teaching Science for Understanding James Joseph Gallagher, 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities

Electrons, Atoms, and Molecules in Inorganic Chemistry Joseph J. Stephanos, Anthony W. Addison, 2017-06-01 Electrons Atoms and Molecules in Inorganic Chemistry A Worked Examples Approach builds from fundamental units into molecules to provide the reader with a full understanding of inorganic chemistry concepts through worked examples and full color illustrations The book uniquely discusses failures as well as research success stories Worked problems include a variety of types of chemical and physical data illustrating the interdependence of issues This text contains a bibliography providing access to important review articles and papers of relevance as well as summaries of leading articles and reviews at the end of each chapter so interested readers can readily consult the original literature Suitable as a professional reference for researchers in a variety of fields as well as course use and self study The book offers valuable information to fill an important gap in the field Incorporates questions and answers to assist readers in understanding a variety of problem types Includes detailed explanations and developed practical approaches for solving real chemical problems Includes a range of example levels from classic and simple for basic concepts to complex questions for more sophisticated topics Covers the full range of topics in inorganic chemistry electrons and wave particle duality electrons in atoms chemical binding molecular symmetry theories of bonding valence bond theory VSEPR theory orbital hybridization molecular orbital theory crystal field theory ligand field theory electronic spectroscopy vibrational and rotational spectroscopy

Prentice Hall Chemistry Antony C. Wilbraham, 2008 *Chemistry* Pearson

Education, Lemay, 2005-05 This lecture notebook contains the art from the text with note taking sections to obviate the need for students to spend time redrawing figures in lecture and instead lets them concentrate on taking notes

Comprehensive Inorganic Chemistry II, 2013-07-23 Comprehensive Inorganic Chemistry II Nine Volume Set reviews and examines topics of relevance to today's inorganic chemists. Covering more interdisciplinary and high impact areas. Comprehensive Inorganic Chemistry II includes biological inorganic chemistry, solid state chemistry, materials chemistry, and nanoscience. The work is designed to follow on with a different viewpoint and format from our 1973 work *Comprehensive Inorganic Chemistry* edited by Bailar, Emelius, Nyholm, and Trotman-Dickenson, which has received over 2,000 citations. The new work will also complement other recent Elsevier works in this area: *Comprehensive Coordination Chemistry* and *Comprehensive Organometallic Chemistry* to form a trio of works covering the whole of modern inorganic chemistry. Chapters are designed to provide a valuable, long-standing scientific resource for both advanced students new to an area and researchers who need further background or answers to a particular problem on the elements, their compounds, or applications. Chapters are written by teams of leading experts under the guidance of the Volume Editors and the Editors in Chief. The articles are written at a level that allows undergraduate students to understand the material while providing active researchers with a ready reference resource for information in the field. The chapters will not provide basic data on the elements, which is available from many sources and the original work, but instead concentrate on applications of the elements and their compounds. Provides a comprehensive review which serves to put many advances in perspective and allows the reader to make connections to related fields such as biological inorganic chemistry, materials chemistry, solid state chemistry, and nanoscience. Inorganic chemistry is rapidly developing, which brings about the need for a reference resource such as this that summarises recent developments and simultaneously provides background information. Forms the new definitive source for researchers interested in elements and their applications, completely replacing the highly cited first edition, which was published in 1973. Study Guide [to Accompany] General Chemistry Dixie Goss, 2002 **Inorganic Fluorine Chemistry**

Joseph S. Thrasher, 1994 Offers an up-to-date account of fundamental and applied research and points to future directions in inorganic fluorine chemistry. Reviews the major advances in fluorine-containing substituent groups. Provides a critical examination of coordination numbers greater than six among fluorides and oxofluorides. Discusses attempts to control chemical reactivity and stability through the use of more nucleophilic sources of fluoride or conversely more weakly coordinating anions. Features chapters on new fluorine-containing ligands in organometallic, transition metal, solid state, lanthanide, and actinide chemistry. Powerpoint Lecture Notebook Theodore Brown, Jr., H. LeMay, Bruce Bursten, 2002-08

The Group 13 Metals: Aluminium, Gallium, Indium, and Thallium Simon Aldridge, Anthony J. Downs, 2011-04-11 The last two decades have seen a renaissance in interest in the chemistry of the main group elements. In particular, research on the metals of group 13, aluminium, gallium, indium, and thallium, has led to the synthesis and isolation of some very novel and

unusual molecules with implications for organometallic synthesis new materials development and with biological medical and environmental relevance The Group 13 Metals Aluminium Gallium Indium and Thallium aims to cover new facts developments and applications in the context of more general patterns of physical and chemical behaviour Particular attention is paid to the main growth areas including the chemistry of lower formal oxidation states cluster chemistry the investigation of solid oxides and hydroxides advances in the formation of III V and related compounds the biological significance of Group 13 metal complexes and the growing importance of the metals and their compounds in the mediation of organic reactions Chapters cover general features of the group 13 elements group 13 metals in the 3 oxidation state simple inorganic compounds formal oxidation state 3 organometallic chemistry formal oxidation state 2 metal metal bonded vs mononuclear derivatives group 13 metals in the 1 oxidation state mixed or intermediate valence group 13 metal compounds aluminium and gallium clusters metalloid clusters and their relation to the bulk phases to naked clusters and to nanoscaled materials simple and mixed metal oxides and hydroxides solids with extended structures of different dimensionalities and porosities coordination and solution chemistry of the metals biological medical and environmental relevance III V and related semiconductor materials group 13 metal mediated organic reactions The Group 13 Metals Aluminium Gallium Indium and Thallium provides a detailed wide ranging and up to date review of the chemistry of this important group of metals It will find a place on the bookshelves of practitioners researchers and students working in inorganic organometallic and materials chemistry

Elements of Inorganic Structural Chemistry Erwin Parthé,1996

Fundamentals of Water Treatment Unit

Processes David Hendricks,2010-11-09 Carefully designed to balance coverage of theoretical and practical principles Fundamentals of Water Treatment Unit Processes delineates the principles that support practice using the unit processes approach as the organizing concept The author covers principles common to any kind of water treatment for example drinking water municipal wastewater industrial water treatment industrial waste water treatment and hazardous wastes Since technologies change but principles remain constant the book identifies strands of theory rather than discusses the latest technologies giving students a clear understanding of basic principles they can take forward in their studies Reviewing the historical development of the field and highlighting key concepts for each unit process each chapter follows a general format that consists of process description history theory practice problems references and a glossary This organizational style facilitates finding sections of immediate interest without having to page through an excessive amount of material Pedagogical Features End of chapter glossaries provide a ready reference and add terms pertinent to topic but beyond the scope of the chapter Sidebars sprinkled throughout the chapters present the lore and history of a topic enlarging students perspective Example problems emphasize tradeoffs and scenarios rather than single answers and involve spreadsheets Reference material includes several appendices and a quick reference spreadsheet Solutions manual includes spreadsheets for problems Supporting material is available for download Understanding how the field arrived at its present state of the art

places the technology in a more logical context and gives students a strong foundation in basic principles This book does more than build technical proficiency it adds insight and understanding to the broader aspects of water treatment unit processes

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