

Name :

Date :

# Chemistry Atomic Structure Worksheet

① For the atomic structure shown below right, answer the following questions.

a) Which element is the atom shown above? \_\_\_\_\_

b) What is the chemical symbol of the element? \_\_\_\_\_

c) What is the mass of the atom shown above? \_\_\_\_\_

d) How many valence electrons are shown in the atom above? \_\_\_\_\_



② Which subatomic particle did J.J. Thomson discover in order to create his plum pudding model of the atom? \_\_\_\_\_

③ The nucleus was discovered by which scientist:

a) Thomson

b) Dalton

c) Rutherford

d) Democritus

④ What is the mass of Carbon-14: \_\_\_\_\_

⑤ What is the number of protons in Carbon-14: \_\_\_\_\_

⑥ What is the number of neutrons in Carbon-14: \_\_\_\_\_

⑦ What is the number of protons in the radioactive Uranium-238: \_\_\_\_\_

⑧ What is the number of neutrons in Sulfur-33: \_\_\_\_\_

⑨ Neutral atoms of the same element can differ in

a) Mass number

b) Atomic number

c) Number of protons and electrons

⑩ The basic principles of atomic theory were first conceived by

a) Avogadro

b) Bohr

c) Dalton

# Pearson Chemistry Atomic Structure Test Answers

**Rosina Ehmann**



**Pearson Chemistry Atomic Structure Test Answers:**

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In a digitally-driven world where screens reign supreme and quick interaction drowns out the subtleties of language, the profound strategies and mental nuances hidden within phrases frequently move unheard. Yet, located within the pages of **Pearson Chemistry Atomic Structure Test Answers** a captivating fictional treasure pulsating with organic thoughts, lies an extraordinary quest waiting to be undertaken. Published by a talented wordsmith, that charming opus encourages viewers on an introspective journey, gently unraveling the veiled truths and profound affect resonating within the very material of each word. Within the mental depths with this emotional evaluation, we will embark upon a honest exploration of the book is core themes, dissect their captivating writing style, and succumb to the strong resonance it evokes serious within the recesses of readers hearts.

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